Chemistry of individual molecules through a lens of transmission electron microscope

A.N. Khlobystov¹, T. Zoberbier², T. W. Chamberlain¹, J. Biskupek², S. T. Skowron¹, E. Bichoutskaia¹, U. Kaiser²

- School of Chemistry, University of Nottingham, University Park, NG7 2RD United Kingdom
 Electron Microscopy of Materials Science, Central Facility for Electron Microscopy, Ulm University, Albert Einstein Allee 11, Ulm 89081, Germany

E-mail: Andrei.Khlobystov@nottingham.ac.uk

Transmission electron microscopy (TEM) is one of the most direct methods for imaging individual molecules with atomic resolution, in direct space and real time. The latest achievements in highresolution TEM not only provide exquisite structural information, but also help to answer some of the most fundamental and challenging questions of chemical science.

Fast electrons employed for TEM imaging transfer some of their kinetic energy to the specimen, which is traditionally viewed as detrimental to the analysis. However, if atomistic mechanisms of interactions between fast electrons and molecules are understood, we can embrace the changes in materials triggered by the act of observation to reveal valuable chemical information. The concept of the electron beam simultaneously acting as an imaging tool and a source of energy to drive chemical transformations offers an entirely new tool for studying chemical reactions of individual molecules with atomic resolution.

Atomically thin and structurally robust carbon nanotubes serve as ultimate nanoscale test tubes, providing an imaging platform for encapsulated molecules. The electron beam of TEM penetrates the walls of carbon nanotubes and enables time-resolved imaging of the reactions between the molecules, which can be followed by TEM with sub-Angstrom resolution. Activated by the electron beam, the energy and dose rate of which can be set precisely, molecules undergo different chemical transformations determined by the imaging conditions. Some types of the reactions revealed under the e-beam are familiar to chemists (i.e. polymerisation) and their mechanisms can be studied and interpreted with the atomic precision, while others are newly discovered by TEM thus changing the way chemists make and study molecules.

Acknowledgement

This work is supported by the Engineering & Physical Sciences Research Council (EPSRC), European Research Council (ERC), German Research Foundation (DFG) and the Ministry of Science, Research and Arts (MWK) of the state Baden- Wurttemberg within the Sub-Ångstrøm Low-Voltage Electron Microscopy project (SALVE).

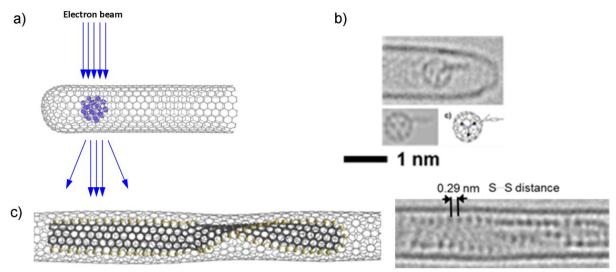


Fig. 1. (a) A single-walled carbon nanotube serves as a container for molecules (b) which can be imaged by TEM with atomic precision. Activated by the electron beam the molecules undergo chemical transformations which can be followed by time-resolved TEM imaging, and in some cases lead to new, unexpected products such as (c) a graphene nanoribbon.

.